

**Lampiran 1. Penjabaran Persamaan (28) menjadi (29)**

$$\begin{aligned} C &= \int_0^{t_3-t_1} A dt \\ &= As(1-e^{-\lambda t_1})(e^{-\lambda(t_2-t_1)}) \int_0^{t_3-t_2} e^{-\lambda t} dt \\ &= \frac{As}{\lambda} (1-e^{-\lambda t_1})(e^{-\lambda(t_2-t_1)})(1-e^{-\lambda(t_3-t_2)}) \end{aligned}$$

$$As = \frac{\lambda C}{(1-e^{-\lambda t_1})(e^{-\lambda(t_2-t_1)})(1-e^{-\lambda(t_3-t_2)})}$$

**Lampiran 2. Penjabaran Persamaan (30) menjadi (31)**

$$k\Sigma_{act}\phi V = \frac{\lambda C}{(1-e^{-\lambda t_1})(e^{-\lambda(t_2-t_1)})(1-e^{-\lambda(t_3-t_2)})}$$

$$\phi = \frac{\lambda C}{kV\Sigma_{act}(1-e^{-\lambda t_1})(e^{-\lambda(t_2-t_1)})(1-e^{-\lambda(t_3-t_2)})}$$

$$\phi = \frac{\lambda C}{k \frac{Bc}{\rho} \sum_{act} (1-e^{-\lambda t_1})(e^{-\lambda t_2-t_1})(1-e^{-\lambda t_3-t_2})}$$

$$\phi = \frac{\lambda C \rho}{kBc \sum_{act} (1-e^{-\lambda t_1})(e^{-\lambda t_2-t_1})(1-e^{-\lambda t_3-t_2})}$$

### Lampiran 3. Penjabaran Persamaan (32)

$$\begin{aligned}\phi &= \frac{As}{k\Sigma_{act}V} \\ &= \frac{As}{kN\sigma_{act}V} \\ &= \frac{As}{k\frac{\rho}{A}N\sigma_{act}V} \\ \phi &= \frac{AsA}{kN_A\sigma_{act}Bc}\end{aligned}$$

#### Lampiran 4. Penjabaran persamaan (33)

Dari persamaan (31) bahwa  $\phi = \frac{\lambda C \rho}{k B_c \sum_{act} (1 - e^{-\lambda t_i})(e^{-\lambda t_2 - t_1})(1 - e^{-\lambda t_3 - t_2})}$  dan

Persamaan (33) bahwa  $\Delta \phi = \left| \frac{\partial \phi}{\partial B_c} \right| \Delta B_c + \left| \frac{\partial \phi}{\partial t_i} \right| \Delta t_i + \left| \frac{\partial \phi}{\partial t_d} \right| \Delta t_d + \left| \frac{\partial \phi}{\partial t_c} \right| \Delta t_c$

Diperoleh turunan dari fluks neutron terhadap Bc dan As sebagai berikut:

$$\frac{\partial \phi}{\partial B_c} = - \frac{\lambda C \rho}{k \sum_{act} B_c^2 (1 - e^{-\lambda t_i})(e^{-\lambda t_d})(1 - e^{-\lambda t_c})}$$

$$\frac{\partial \phi}{\partial t_i} = - \frac{\lambda^2 C \rho (e^{-\lambda t_i})}{k \sum_{act} B_c (1 - e^{-\lambda t_i})^2 (e^{-\lambda t_d})(1 - e^{-\lambda t_c})}$$

$$\frac{\partial \phi}{\partial t_d} = \frac{\lambda^2 C \rho (e^{-\lambda t_d})}{k \sum_{act} B_c (1 - e^{-\lambda t_i})(e^{-\lambda t_d})(1 - e^{-\lambda t_c})}$$

$$\frac{\partial \phi}{\partial t_c} = - \frac{\lambda^2 C \rho (e^{-\lambda t_c})}{k \sum_{act} B_c (1 - e^{-\lambda t_i})(e^{-\lambda t_d})(1 - e^{-\lambda t_c})^2}$$

Turunan diatas disubstitusikan ke persamaan (33)

$$\begin{aligned} \Delta \phi &= \left| - \frac{\lambda C \rho}{k \sum_{act} B_c^2 (1 - e^{-\lambda t_i})(e^{-\lambda t_d})(1 - e^{-\lambda t_c})} \right| \Delta B_c + \left| - \frac{\lambda^2 C \rho (e^{-\lambda t_i})}{k \sum_{act} B_c (1 - e^{-\lambda t_i})^2 (e^{-\lambda t_d})(1 - e^{-\lambda t_c})} \right| \Delta t_i \\ &+ \left| \frac{\lambda^2 C \rho (e^{-\lambda t_d})}{k \sum_{act} B_c (1 - e^{-\lambda t_i})(e^{-\lambda t_d})(1 - e^{-\lambda t_c})} \right| \Delta t_d + \left| - \frac{\lambda^2 C \rho (e^{-\lambda t_c})}{k \sum_{act} B_c (1 - e^{-\lambda t_i})(e^{-\lambda t_d})(1 - e^{-\lambda t_c})^2} \right| \Delta t_c \\ &= \frac{\lambda^2 C \rho}{k \sum_{act} B_c (1 - e^{-\lambda t_i})(e^{-\lambda t_d})(1 - e^{-\lambda t_c})} \left[ \frac{1}{\lambda B_c} \Delta B_c + \frac{e^{-\lambda t_i}}{1 - e^{-\lambda t_i}} \Delta t_i + \Delta t_d + \frac{e^{-\lambda t_c}}{1 - e^{-\lambda t_c}} \Delta t_c \right] \end{aligned}$$

### Lampiran 5. Penjabaran Persamaan (36)

Dari persamaan (34)  $CR = \frac{As}{As_{cd}}$

$$As_{cd} = \frac{As}{CR}$$

Fluks neutron thermal ditentukan dari hubungan:

$$\phi = c(As - As_{cd})$$

$$= c \left[ As - \frac{As}{CR} \right]$$

$$= cAs - \frac{cAs}{CR}$$

$$= cAs \left[ 1 - \frac{1}{CR} \right]$$

$$\phi = cAs \left[ \frac{CR-1}{CR} \right]$$

## Lampiran 6. Penjabaran Persamaan (52)

Fluks neutron thermal adalah selisih antara fluks neutron total dan neutron cepat:

$$\phi_{th} = \phi_{\tau} - \phi_c$$

dengan

$$\phi_{\tau} = \phi_{\tau} \pm \phi_{\tau}$$

$$\phi_c = \phi_c \pm \phi_c$$

sehingga

$$\begin{aligned}\phi_{th} &= \left| \frac{\partial \phi_{th}}{\partial \phi_{\tau}} \right| \Delta \phi_{\tau} + \left| \frac{\partial \phi_{th}}{\partial \phi_c} \right| \Delta \phi_c \\ &= |1| \Delta \phi_{\tau} + |-1| \Delta \phi_c \\ &= \phi_{\tau} + \phi_c\end{aligned}$$

### Lampiran 7. Penjabaran Persamaan (53)

$$R_{cd} = \frac{\phi_\tau}{\phi_c}$$

$$\Delta R_{cd} = \left| \frac{\partial R_{cd}}{\partial \phi_\tau} \right| \Delta \phi_\tau + \left| \frac{\partial R_{cd}}{\partial \phi_c} \right| \Delta \phi_c$$

$$\Delta R_{cd} = |\phi_c^{-1}| \Delta \phi_\tau + |-\phi_\tau \phi_c^{-2}| \Delta \phi_c$$

$$\Delta R_{cd} = \left| \frac{1}{\phi_c} \right| \Delta \phi_\tau + \left| \frac{-\phi_\tau}{\phi_c^2} \right| \Delta \phi_c$$

## Lampiran 8. Perhitungan Aktivitas pada Kalibrasi Efisiensi Detektor

Besarnya Aktivitas :

$$A = A_0 e^{-\lambda t}$$

dengan :

$$A_0 = 1,975 \times 10^5$$

Waktu awal = 15 Juni 1979

Waktu Pencacahan = 25 April 2012

Waktu paro ( $T_{1/2}$ ) = 12,7 tahun

Waktu tunda( $t$ ) = 32,861 tahun

$$\lambda = \frac{0,693}{T_{1/2}} = 5,4567 \times 10^{-2} \text{ tahun}^{-1}$$

Maka dapat dihitung besarnya aktivitas sebagai berikut:

$$A = A_0 e^{-\lambda t}$$

$$A = 1,975 \times 10^5 e^{-(5,4567 \times 10^{-2})(32,861)}$$

$$= 32872,40363 \text{ dps}$$



## Lampiran 9. Perhitungan Efisiensi Detektor HPGe dengan Sumber

Standar  $^{152}_{63}\text{Eu}$

$$\% k(E) = \frac{\text{cps}}{\text{dps} \cdot Y(E)} \times 100\%$$

$$1. \quad k(121,77) = \frac{149,32}{32872,404 (0,2924)} = 0,01553492 = 1,55 \%$$

$$2. \quad k(244,59) = \frac{26,32}{32872,404 (0,0761)} = 0,01052131 = 1,05 \%$$

$$3. \quad k(344,21) = \frac{64,76}{32872,404 (0,2700)} = 0,00729645 = 0,73 \%$$

$$4. \quad k(443,97) = \frac{3,34}{32872,404 (0,0284)} = 0,00357764 = 0,36 \%$$

$$5. \quad k(778,93) = \frac{13,01}{32872,404 (0,1299)} = 0,00304675 = 0,30 \%$$

$$6. \quad k(964,05) = \frac{11,37}{32872,404 (0,1458)} = 0,00237231 = 0,24 \%$$

$$7. \quad k(1112,05) = \frac{8,99}{32872,404 (0,1358)} = 0,00201386 = 0,20 \%$$

$$8. \quad k(1407,87) = \frac{11,13}{32872,404 (0,2121)} = 0,00159633 = 0,16 \%$$

## Lampiran 10. Perhitungan Nilai Fluks Neutron

Dengan mengetahui hasil cacah dari detektor dan data-data lainnya, dapat dihitung besarnya aktivitas dan fluks neutron dengan menggunakan persamaan (29) dan (32)

1. Data  $^{197}_{79}\text{Au}$  foil yang tidak dibungkus cadmium (Fluks Neutron Total)

a. Data cuplikan :

Cuplikan = $^{197}_{79}\text{Au}$	$t_i = 600$ sekon
A = 197	$t_d = 695340$ sekon
Bc = 0,004 gram	$t_c = 600$ sekon
$\sigma_{act} = 98,8$ barn = $9,88 \times 10^{-23} \text{ cm}^2$	C = 4900,71 cps
$\lambda = 2,9762 \times 10^{-6} \text{ s}^{-1}$	k = $0,00554 = 5,54 \times 10^{-3}$
$N_A = 6,023 \times 10^{23}$ atom/mol	

$$A_s = \frac{\lambda C}{(1 - e^{-\lambda t_1})(e^{-\lambda(t_2 - t_1)})(1 - e^{-\lambda(t_3 - t_2)})}$$

$$A_s = \frac{(2,9762 \times 10^{-6})(4900,71 \times 600)}{(1 - e^{-(2,9762 \times 10^{-6})(600)})(e^{-(2,9762 \times 10^{-6})(695340)})(1 - e^{-(2,9762 \times 10^{-6})(600)})}$$

$$= 21776137,49 \text{ dps}$$

$$\phi = \frac{A_s A}{K N_A \sigma_{act} Bc}$$

$$\phi = \frac{(21776137,49)(197)}{(5,54 \times 10^{-3})(6,023 \times 10^{23})(9,88 \times 10^{-23})(0,004)}$$

$$\phi = 3,25 \times 10^{12} \text{ ncm}^{-2} \text{ s}^{-1}$$

b. Data cuplikan :

$$\begin{aligned}
 \text{Cuplikan} &= {}^{197}_{79}\text{Au} & t_i &= 600 \text{ sekon} \\
 A &= 197 & t_d &= 696960 \text{ sekon} \\
 Bc &= 0,005 \text{ gram} & t_c &= 600 \text{ sekon} \\
 \sigma_{act} &= 98,8 \text{ barn} = 9,88 \times 10^{-23} \text{ cm}^2 & C &= 5437,63 \text{ cps} \\
 \lambda &= 2,9762 \times 10^{-6} \text{ s}^{-1} & k &= 0,00554 = 5,54 \times 10^{-3} \\
 N_A &= 6,023 \times 10^{23} \text{ atom/mol}
 \end{aligned}$$

$$A_s = \frac{\lambda C}{(1 - e^{-\lambda t_1})(e^{-\lambda(t_2 - t_1)})(1 - e^{-\lambda(t_3 - t_2)})}$$

$$A_s = \frac{(2,9762 \times 10^{-6})(5437,63 \times 600)}{(1 - e^{-(2,9762 \times 10^{-6})(600)})(e^{-(2,9762 \times 10^{-6})(696960)})(1 - e^{-(2,9762 \times 10^{-6})(600)})}$$

$$= 24278699,82 \text{ dps}$$

$$\phi = \frac{AsA}{KN_A \sigma_{act} Bc}$$

$$\phi = \frac{(24278699,82)(197)}{(5,54 \times 10^{-3})(6,023 \times 10^{23})(9,88 \times 10^{-23})(0,005)}$$

$$\phi = 2,90 \times 10^{12} \text{ ncm}^{-2} \text{ s}^{-1}$$

c. Data cuplikan :

$$\begin{aligned}
 \text{Cuplikan} &= {}^{197}_{79}\text{Au} & t_i &= 600 \text{ sekon} \\
 A &= 197 & t_d &= 706200 \text{ sekon} \\
 Bc &= 0,006 \text{ gram} & t_c &= 600 \text{ sekon} \\
 \sigma_{act} &= 98,8 \text{ barn} = 9,88 \times 10^{-23} \text{ cm}^2 & C &= 3409,51 \text{ cps} \\
 \lambda &= 2,9762 \times 10^{-6} \text{ s}^{-1} & k &= 0,00554 = 5,54 \times 10^{-3}
 \end{aligned}$$

$$N_A = 6,023 \times 10^{23} \text{ atom/mol}$$

$$A_s = \frac{\lambda C}{(1-e^{-\lambda t_1})(e^{-\lambda(t_2-t_1)})(1-e^{-\lambda(t_3-t_2)})}$$

$$A_s = \frac{(2,9762 \times 10^{-6})(3409,51 \times 600)}{(1-e^{-(2,9762 \times 10^{-6})(600)})(e^{-(2,9762 \times 10^{-6})(706200)})(1-e^{-(2,9762 \times 10^{-6})(600)})}$$

$$= 15647713,20 \text{ dps}$$

$$\phi = \frac{AsA}{KN_A \sigma_{act} Bc}$$

$$\phi = \frac{(15647713,20)(197)}{(5,54 \times 10^{-3})(6,023 \times 10^{23})(9,88 \times 10^{-23})(0,006)}$$

$$\phi = 1,56 \times 10^{12} \text{ ncm}^{-2} \text{ s}^{-1}$$

d. Data cuplikan :

$$\text{Cuplikan} = {}^{197}_{79}\text{Au}$$

$$t_i = 600 \text{ sekon}$$

$$A = 197$$

$$t_d = 707580 \text{ sekon}$$

$$Bc = 0,007 \text{ gram}$$

$$t_c = 600 \text{ sekon}$$

$$\sigma_{act} = 98,8 \text{ barn} = 9,88 \times 10^{-23} \text{ cm}^2$$

$$C = 2203,84 \text{ cps}$$

$$\lambda = 2,9762 \times 10^{-6} \text{ s}^{-1}$$

$$k = 0,00554 = 5,54 \times 10^{-3}$$

$$N_A = 6,023 \times 10^{23} \text{ atom/mol}$$

$$A_s = \frac{\lambda C}{(1-e^{-\lambda t_1})(e^{-\lambda(t_2-t_1)})(1-e^{-\lambda(t_3-t_2)})}$$

$$A_s = \frac{(2,9762 \times 10^{-6})(2203,84 \times 600)}{(1-e^{-(2,9762 \times 10^{-6})(600)})(e^{-(2,9762 \times 10^{-6})(707580)})(1-e^{-(2,9762 \times 10^{-6})(600)})}$$

$$= 10155999,84 \text{ dps}$$

$$\phi = \frac{AsA}{KN_A\sigma_{act}Bc}$$

$$\phi = \frac{(10155999,84)(197)}{(5,54 \times 10^{-3})(6,023 \times 10^{23})(9,88 \times 10^{-23})(0,007)}$$

$$\phi = 8,69 \times 10^{11} \text{ncm}^{-2}\text{s}^{-1}$$

e. Data cuplikan :

$$\text{Cuplikan} = {}^{197}_{79}\text{Au}$$

$$t_i = 600 \text{ sekon}$$

$$A = 197$$

$$t_d = 709020 \text{ sekon}$$

$$Bc = 0,009 \text{ gram}$$

$$t_c = 600 \text{ sekon}$$

$$\sigma_{act} = 98,8 \text{ barn} = 9,88 \times 10^{-23} \text{ cm}^2$$

$$C = 584,53 \text{ cps}$$

$$\lambda = 2,9762 \times 10^{-6} \text{ s}^{-1}$$

$$k = 0,00554 = 5,54 \times 10^{-3}$$

$$N_A = 6,023 \times 10^{23} \text{ atom/mol}$$

$$A_s = \frac{\lambda C}{(1 - e^{-\lambda t_1})(e^{-\lambda(t_2 - t_1)})(1 - e^{-\lambda(t_3 - t_2)})}$$

$$A_s = \frac{(2,9762 \times 10^{-6})(584,53 \times 600)}{(1 - e^{-(2,9762 \times 10^{-6})(600)})(e^{-(2,9762 \times 10^{-6})(709020)})(1 - e^{-(2,9762 \times 10^{-6})(600)})}$$

$$= 2705270,51 \text{ dps}$$

$$\phi = \frac{AsA}{KN_A\sigma_{act}Bc}$$

$$\phi = \frac{(2705270,51)(197)}{(5,54 \times 10^{-3})(6,023 \times 10^{23})(9,88 \times 10^{-23})(0,009)}$$

$$\phi = 1,80 \times 10^{11} \text{ncm}^{-2}\text{s}^{-1}$$

2. Data foil  $^{197}_{79}\text{Au}$  yang dibungkus cadmium (Fluks Neutron Cepat)

a. Data cuplikan :

Cuplikan	$= ^{197}_{79}\text{Au}$	$t_i = 600$ sekon
A	$= 197$	$t_d = 696240$ sekon
Bc	$= 0,004$ gram	$t_c = 600$ sekon
$\sigma_{act}$	$= 98,8$ barn $= 9,88 \times 10^{-23} \text{ cm}^2$	$C = 1597,51 \text{ cps}$
$\lambda$	$= 2,9762 \times 10^{-6} \text{ s}^{-1}$	$k = 0,00553 = 5,53 \times 10^{-3}$
$N_A$	$= 6,023 \times 10^{23}$ atom/mol	

$$A_s = \frac{\lambda C}{(1 - e^{-\lambda t_1})(e^{-\lambda(t_2 - t_1)})(1 - e^{-\lambda(t_3 - t_2)})}$$

$$A_s = \frac{(2,9762 \times 10^{-6})(1597,51 \times 600)}{(1 - e^{-(2,9762 \times 10^{-6})(600)})(e^{-(2,9762 \times 10^{-6})(696240)})(1 - e^{-(2,9762 \times 10^{-6})(600)})}$$

$$= 7117654,12 \text{ dps}$$

$$\phi = \frac{AsA}{KN_A \sigma_{act} Bc}$$

$$\phi = \frac{(7117654,12)(197)}{(5,53 \times 10^{-3})(6,023 \times 10^{23})(9,88 \times 10^{-23})(0,004)}$$

$$\phi = 1,06 \times 10^{12} \text{ ncm}^{-2} \text{ s}^{-1}$$

b. Data cuplikan :

Cuplikan	$= ^{197}_{79}\text{Au}$	$t_i = 600$ sekon
A	$= 197$	$t_d = 697920$ sekon
Bc	$= 0,005$ gram	$t_c = 600$ sekon
$\sigma_{act}$	$= 98,8$ barn $= 9,88 \times 10^{-23} \text{ cm}^2$	$C = 1494,24 \text{ cps}$

$$\lambda = 2,9762 \times 10^{-6} \text{ s}^{-1} \quad k = 0,00553 = 5,53 \times 10^{-3}$$

$$N_A = 6,023 \times 10^{23} \text{ atom/mol}$$

$$A_s = \frac{\lambda C}{(1 - e^{-\lambda t_1})(e^{-\lambda(t_2 - t_1)})(1 - e^{-\lambda(t_3 - t_2)})}$$

$$A_s = \frac{(2,9762 \times 10^{-6})(1494,24 \times 600)}{(1 - e^{-(2,9762 \times 10^{-6})(600)})(e^{-(2,9762 \times 10^{-6})(697920)})(1 - e^{-(2,9762 \times 10^{-6})(600)})}$$

$$= 6690783,48 \text{ dps}$$

$$\phi = \frac{AsA}{KN_A \sigma_{act} Bc}$$

$$\phi = \frac{(6690783,48)(197)}{(5,53 \times 10^{-3})(6,023 \times 10^{23})(9,88 \times 10^{-23})(0,005)}$$

$$\phi = 7,99 \times 10^{11} \text{ ncm}^{-2} \text{ s}^{-1}$$

c. Data cuplikan :

$$\text{Cuplikan} = {}^{197}_{79}\text{Au} \quad t_i = 600 \text{ sekon}$$

$$A = 197 \quad t_d = 706980 \text{ sekon}$$

$$Bc = 0,006 \text{ gram} \quad t_c = 600 \text{ sekon}$$

$$\sigma_{act} = 98,8 \text{ barn} = 9,88 \times 10^{-23} \text{ cm}^2 \quad C = 433,43 \text{ cps}$$

$$\lambda = 2,9762 \times 10^{-6} \text{ s}^{-1} \quad k = 0,00553 = 5,53 \times 10^{-3}$$

$$N_A = 6,023 \times 10^{23} \text{ atom/mol}$$

$$A_s = \frac{\lambda C}{(1 - e^{-\lambda t_1})(e^{-\lambda(t_2 - t_1)})(1 - e^{-\lambda(t_3 - t_2)})}$$

$$A_s = \frac{(2,9762 \times 10^{-6})(433,43 \times 600)}{(1 - e^{-(2,9762 \times 10^{-6})(600)})(e^{-(2,9762 \times 10^{-6})(706980)})(1 - e^{-(2,9762 \times 10^{-6})(600)})}$$

$$= 1993820,53 \text{ dps}$$

$$\phi = \frac{AsA}{KN_A \sigma_{act} Bc}$$

$$\phi = \frac{(1993820,53)(197)}{(5,53 \times 10^{-3})(6,023 \times 10^{23})(9,88 \times 10^{-23})(0,006)}$$

$$\phi = 1,99 \times 10^{11} \text{ ncm}^{-2} \text{ s}^{-1}$$

d. Data cuplikan :

Cuplikan	= $^{197}_{79}\text{Au}$	$t_i = 600$ sekon
A	= 197	$t_d = 708300$ sekon
Bc	= 0,007 gram	$t_c = 600$ sekon
$\sigma_{act}$	= 98,8 barn = $9,88 \times 10^{-23} \text{ cm}^2$	C = 220,22cps
$\lambda$	= $2,9762 \times 10^{-6} \text{ s}^{-1}$	k = 0,00553 = $5,53 \times 10^{-3}$
$N_A$	= $6,023 \times 10^{23}$ atom/mol	

$$A_s = \frac{\lambda C}{(1 - e^{-\lambda t_1})(e^{-\lambda(t_2 - t_1)})(1 - e^{-\lambda(t_3 - t_2)})}$$

$$A_s = \frac{(2,9762 \times 10^{-6})(220,22 \times 600)}{(1 - e^{-(2,9762 \times 10^{-6})(600)})(e^{-(2,9762 \times 10^{-6})(708300)})(1 - e^{-(2,9762 \times 10^{-6})(600)})}$$

$$= 1017021,22 \text{ dps}$$

$$\phi = \frac{AsA}{KN_A \sigma_{act} Bc}$$



$$\phi = \frac{(1017021,22)(197)}{(5,53 \times 10^{-3})(6,023 \times 10^{23})(9,88 \times 10^{-23})(0,007)}$$

$$\phi = 8,71 \times 10^{10} \text{ncm}^{-2} \text{s}^{-1}$$

e. Data cuplikan :

$$\text{Cuplikan} = {}^{197}_{79}\text{Au}$$

$$t_i = 600 \text{ sekon}$$

$$A = 197$$

$$t_d = 709680 \text{ sekon}$$

$$Bc = 0,009 \text{ gram}$$

$$t_c = 600 \text{ sekon}$$

$$\sigma_{act} = 98,8 \text{ barn} = 9,88 \times 10^{-23} \text{ cm}^2$$

$$C = 26,54 \text{ cps}$$

$$\lambda = 2,9762 \times 10^{-6} \text{ s}^{-1}$$

$$k = 0,00553 = 5,53 \times 10^{-3}$$

$$N_A = 6,023 \times 10^{23} \text{ atom/mol}$$

$$A_s = \frac{\lambda C}{(1 - e^{-\lambda t_1})(e^{-\lambda(t_2 - t_1)})(1 - e^{-\lambda(t_3 - t_2)})}$$

$$A_s = \frac{(2,9762 \times 10^{-6})(26,54 \times 600)}{(1 - e^{-(2,9762 \times 10^{-6})(600)})(e^{-(2,9762 \times 10^{-6})(709680)}) (1 - e^{-(2,9762 \times 10^{-6})(600)})}$$

$$= 123071,61 \text{ dps}$$

$$\phi = \frac{AsA}{KN_A \sigma_{act} Bc}$$

$$\phi = \frac{(123071,61)(197)}{(5,53 \times 10^{-3})(6,023 \times 10^{23})(9,88 \times 10^{-23})(0,009)}$$

$$\phi = 8,19 \times 10^9 \text{ncm}^{-2} \text{s}^{-1}$$

### Lampiran 11. Perhitungan Nilai Ralat Fluks Neutron

1. Data  $^{197}_{79}\text{Au}$  foil yang tidak dibungkus cadmium (Fluks Neutron Total)

a. Data cuplikan :

Cuplikan	= $^{197}_{79}\text{Au}$	$t_i$	= 600 sekon
$\rho$	= 19,32 gram $\text{cm}^{-3}$	$t_d$	= 695340 sekon
Bc	= 0,004 gram	$t_c$	= 600 sekon
$\sigma_{act}$	= 98,8 barn = $9,88 \times 10^{-23} \text{cm}^2$	C	= 4900,71 cps
$\lambda$	= $2,9762 \times 10^{-6} \text{s}^{-1}$	k	= 0,00554 = $5,54 \times 10^{-3}$
$N_A$	= $6,023 \times 10^{23}$ atom/mol	$\Sigma_{act}$	= 59,50724
$\Delta t_i = \Delta t_d = \Delta t_c$	= 0,5 sekon	$\Delta B$	= 0,0005 gram

$$\Delta\phi = \frac{\lambda^2 C \rho}{k \sum_{act} B_c (1 - e^{-\lambda t_i})(e^{-\lambda t_d})(1 - e^{-\lambda t_c})} \left[ \frac{1}{\lambda Bc} \Delta Bc + \frac{e^{-\lambda t_i}}{1 - e^{-\lambda t_i}} \Delta t_i + \Delta t_d + \frac{e^{-\lambda t_c}}{1 - e^{-\lambda t_c}} \Delta t_c \right]$$

$$= \left[ \frac{(2,9762 \cdot 10^{-6})^2 (4900,71) (19,32)}{(5,54 \cdot 10^{-3}) (59,50724) (0,004) (1 - e^{-(2,9762 \cdot 10^{-6}) (600)}) (e^{-(2,9762 \cdot 10^{-6}) (695340)}) ((1 - e^{-(2,9762 \cdot 10^{-6}) (600)}) )} \right]$$

$$\left[ \frac{1}{((2,9762 \cdot 10^{-6}) (0,004))} (0,0005) + \frac{e^{-(2,9762 \cdot 10^{-6}) (600)}}{1 - e^{-(2,9762 \cdot 10^{-6}) (600)}} 0,5 + 0,5 + \frac{e^{-(2,9762 \cdot 10^{-6}) (600)}}{1 - e^{-(2,9762 \cdot 10^{-6}) (600)}} 0,5 \right]$$

$$= 6,73 \times 10 \text{ ncm}^{-2} \text{ s}^{-1}$$

b. Data cuplikan :

Cuplikan	= $^{197}_{79}\text{Au}$	$t_i$	= 600 sekon
$\rho$	= 19,32 gram $\text{cm}^{-3}$	$t_d$	= 696960 sekon
Bc	= 0,005 gram	$t_c$	= 600 sekon

$$\begin{array}{ll}
\sigma_{act} & = 98,8 \text{ barn} = 9,88 \times 10^{-23} \text{ cm}^2 & C & = 5437,63 \text{ cps} \\
\lambda & = 2,9762 \times 10^{-6} \text{ s}^{-1} & k & = 0,00554 = 5,54 \times 10^{-3} \\
N_A & = 6,023 \times 10^{23} \text{ atom/mol} & \Sigma_{act} & = 59,50724 \\
\Delta t_i = \Delta t_d = \Delta t_c & = 0,5 \text{ sekon} & \Delta B & = 0,0005 \text{ gram}
\end{array}$$

$$\begin{aligned}
\Delta\phi &= \frac{\lambda^2 C \rho}{k \sum_{act} B_c (1 - e^{-\lambda t_i})(e^{-\lambda t_d})(1 - e^{-\lambda t_c})} \left[ \frac{1}{\lambda Bc} \Delta Bc + \frac{e^{-\lambda t_i}}{1 - e^{-\lambda t_i}} \Delta t_i + \Delta t_d + \frac{e^{-\lambda t_c}}{1 - e^{-\lambda t_c}} \Delta t_c \right] \\
&= \left[ \frac{(2,9762 \times 10^{-6})^2 (5437,63) (19,32)}{(5,54 \times 10^{-3}) (59,50724) (0,005) (1 - e^{-(2,9762 \times 10^{-6})(600)}) (e^{-(2,9762 \times 10^{-6})(600)}) ((1 - e^{-(2,9762 \times 10^{-6})(600)})} )} \right] \\
&\quad \left[ \frac{1}{((2,9762 \times 10^{-6})(0,005))} (0,0005) + \frac{e^{-(2,9762 \times 10^{-6})(600)}}{1 - e^{-(2,9762 \times 10^{-6})(600)}} 0,5 + 0,5 + \frac{e^{-(2,9762 \times 10^{-6})(600)}}{1 - e^{-(2,9762 \times 10^{-6})(600)}} 0,5 \right] \\
&= 4,82 \times 10^7 \text{ ncm}^{-2} \text{ s}^{-1}
\end{aligned}$$

c. Data cuplikan :

$$\begin{array}{ll}
\text{Cuplikan} & = {}^{197}_{79}\text{Au} & t_i & = 600 \text{ sekon} \\
\rho & = 19,32 \text{ gram cm}^{-3} & t_d & = 706200 \text{ sekon} \\
Bc & = 0,006 \text{ gram} & t_c & = 600 \text{ sekon} \\
\sigma_{act} & = 98,8 \text{ barn} = 9,88 \times 10^{-23} \text{ cm}^2 & C & = 3409,51 \text{ cps} \\
\lambda & = 2,9762 \times 10^{-6} \text{ s}^{-1} & k & = 0,00554 = 5,54 \times 10^{-3} \\
N_A & = 6,023 \times 10^{23} \text{ atom/mol} & \Sigma_{act} & = 59,50724 \\
\Delta t_i = \Delta t_d = \Delta t_c & = 0,5 \text{ sekon} & \Delta B & = 0,0005 \text{ gram}
\end{array}$$

$$\Delta\phi = \frac{\lambda^2 C \rho}{k \sum_{act} B_c (1 - e^{-\lambda t_i})(e^{-\lambda t_d})(1 - e^{-\lambda t_c})} \left[ \frac{1}{\lambda B_c} \Delta B_c + \frac{e^{-\lambda t_i}}{1 - e^{-\lambda t_i}} \Delta t_i + \Delta t_d + \frac{e^{-\lambda t_c}}{1 - e^{-\lambda t_c}} \Delta t_c \right]$$

$$= \left[ \frac{(2,9762 \cdot 10^{-6})^2 (3409,51)(19,32)}{(5,54 \cdot 10^{-3})(59,50724)(0,006) \left(1 - e^{-(2,9762 \cdot 10^{-6})(600)}\right) \left(e^{-(2,9762 \cdot 10^{-6})(706200)}\right) \left(1 - e^{-(2,9762 \cdot 10^{-6})(600)}\right)} \right]$$

$$\left[ \frac{1}{((2,9762 \cdot 10^{-6})(0,006))} (0,0005) + \frac{e^{-(2,9762 \cdot 10^{-6})(600)}}{1 - e^{-(2,9762 \cdot 10^{-6})(600)}} 0,5 + 0,5 + \frac{e^{-(2,9762 \cdot 10^{-6})(600)}}{1 - e^{-(2,9762 \cdot 10^{-6})(600)}} 0,5 \right]$$

$$= 2,16 \cdot 10^7 \text{ ncm}^{-2} \text{ s}^{-1}$$

d. Data cuplikan :

Cuplikan	= ${}^{197}_{79}\text{Au}$	$t_i$	= 600 sekon
$\rho$	= 19,32 gram $\text{cm}^{-3}$	$t_d$	= 707580 sekon
Bc	= 0,007 gram	$t_c$	= 600 sekon
$\sigma_{act}$	= 98,8 barn = $9,88 \cdot 10^{-23} \text{ cm}^2$	C	= 2203,84 cps
$\lambda$	= $2,9762 \cdot 10^{-6} \text{ s}^{-1}$	k	= 0,00554 = $5,54 \cdot 10^{-3}$
$N_A$	= $6,023 \cdot 10^{23}$ atom/mol	$\Sigma_{act}$	= 59,50724
$\Delta t_i = \Delta t_d = \Delta t_c$	= 0,5 sekon	$\Delta B$	= 0,0005 gram

$$\Delta\phi = \frac{\lambda^2 C \rho}{k \sum_{act} B_c (1 - e^{-\lambda t_i})(e^{-\lambda t_d})(1 - e^{-\lambda t_c})} \left[ \frac{1}{\lambda B_c} \Delta B_c + \frac{e^{-\lambda t_i}}{1 - e^{-\lambda t_i}} \Delta t_i + \Delta t_d + \frac{e^{-\lambda t_c}}{1 - e^{-\lambda t_c}} \Delta t_c \right]$$

$$= \left[ \frac{(2,9762 \cdot 10^{-6})^2 (2203,84)(19,32)}{(5,54 \cdot 10^{-3})(59,50724)(0,007) \left(1 - e^{-(2,9762 \cdot 10^{-6})(600)}\right) \left(e^{-(2,9762 \cdot 10^{-6})(707580)}\right) \left(1 - e^{-(2,9762 \cdot 10^{-6})(600)}\right)} \right]$$

$$\left[ \frac{1}{((2,9762 \cdot 10^{-6})(0,007))} (0,0005) + \frac{e^{-(2,9762 \cdot 10^{-6})(600)}}{1 - e^{-(2,9762 \cdot 10^{-6})(600)}} 0,5 + 0,5 + \frac{e^{-(2,9762 \cdot 10^{-6})(600)}}{1 - e^{-(2,9762 \cdot 10^{-6})(600)}} 0,5 \right]$$

$$= 1,04 \cdot 10^7 \text{ ncm}^{-2} \text{ s}^{-1}$$

e. Data cuplikan :

Cuplikan	= ${}^{197}_{79}\text{Au}$	$t_i$	= 600 sekon
$\rho$	= 19,32gram $\text{cm}^{-3}$	$t_d$	= 709020 sekon
Bc	= 0,009 gram	$t_c$	= 600 sekon
$\sigma_{act}$	= 98,8 barn = $9,88 \times 10^{-23} \text{cm}^2$	C	= 584,53cps
$\lambda$	= $2,9762 \times 10^{-6} \text{s}^{-1}$	k	= 0,00554 = $5,54 \times 10^{-3}$
$N_A$	= $6,023 \times 10^{23}$ atom/mol	$\Sigma_{act}$	= 59,50724
$\Delta t_i = \Delta t_d = \Delta t_c$	= 0,5 sekon	$\Delta B$	= 0,0005 gram

$$\Delta\phi = \frac{\lambda^2 C \rho}{k \sum_{act} B_c (1 - e^{-\lambda t_i})(e^{-\lambda t_d})(1 - e^{-\lambda t_c})} \left[ \frac{1}{\lambda Bc} \Delta Bc + \frac{e^{-\lambda t_i}}{1 - e^{-\lambda t_i}} \Delta t_i + \Delta t_d + \frac{e^{-\lambda t_c}}{1 - e^{-\lambda t_c}} \Delta t_c \right]$$

$$= \left[ \frac{(2,9762 \cdot 10^{-6})^2 (584,53)(19,32)}{(5,54 \times 10^{-3})(59,50724)(0,009)(1 - e^{-(2,9762 \cdot 10^{-6})(600)})(e^{-(2,9762 \cdot 10^{-6})(709020)})(1 - e^{-(2,9762 \cdot 10^{-6})(600)})} \right]$$

$$\left[ \frac{1}{((2,9762 \cdot 10^{-6})(0,009))} (0,0005) + \frac{e^{-(2,9762 \cdot 10^{-6})(600)}}{1 - e^{-(2,9762 \cdot 10^{-6})(600)}} 0,5 + 0,5 + \frac{e^{-(2,9762 \cdot 10^{-6})(600)}}{1 - e^{-(2,9762 \cdot 10^{-6})(600)}} 0,5 \right]$$

$$= 1,68 \times 10^6 \text{ ncm}^{-2} \text{ s}^{-1}$$

2. Data foil  ${}^{197}_{79}\text{Au}$  yang dibungkus cadmium (Fluks Neutron Cepat)

a. Data cuplikan :

Cuplikan	= ${}^{197}_{79}\text{Au}$	$t_i$	= 600 sekon
$\rho$	= 19,32gram $\text{cm}^{-3}$	$t_d$	= 696240 sekon
Bc	= 0,004 gram	$t_c$	= 600 sekon

$$\begin{array}{llll}
\sigma_{act} & = 98,8 \text{ barn} = 9,88 \times 10^{-23} \text{ cm}^2 & C & = 1587,54 \text{ cps} \\
\lambda & = 2,9762 \times 10^{-6} \text{ s}^{-1} & k & = 0,00553 = 5,5310^{-3} \\
N_A & = 6,023 \times 10^{23} \text{ atom/mol} & \Sigma_{act} & = 59,50724 \\
\Delta t_i = \Delta t_d = \Delta t_c & = 0,5 \text{ sekon} & \Delta B & = 0,0005 \text{ gram}
\end{array}$$

$$\begin{aligned}
\Delta\phi &= \frac{\lambda^2 C \rho}{k \sum_{act} B_c (1 - e^{-\lambda t_i})(e^{-\lambda t_d})(1 - e^{-\lambda t_c})} \left[ \frac{1}{\lambda Bc} \Delta Bc + \frac{e^{-\lambda t_i}}{1 - e^{-\lambda t_i}} \Delta t_i + \Delta t_d + \frac{e^{-\lambda t_c}}{1 - e^{-\lambda t_c}} \Delta t_c \right] \\
&= \left[ \frac{(2,9762 \cdot 10^{-6})^2 (1597,54)(19,32)}{(5,53 \cdot 10^{-3})(59,50724)(0,004) \left(1 - e^{-(2,9762 \cdot 10^{-6})(600)}\right) \left(e^{-(2,9762 \cdot 10^{-6})(696240)}\right) \left(1 - e^{-(2,9762 \cdot 10^{-6})(600)}\right)} \right] \\
&\quad \left[ \frac{1}{((2,9762 \cdot 10^{-6})(0,004))} (0,0005) + \frac{e^{-(2,9762 \cdot 10^{-6})(600)}}{1 - e^{-(2,9762 \cdot 10^{-6})(600)}} 0,5 + 0,5 + \frac{e^{-(2,9762 \cdot 10^{-6})(600)}}{1 - e^{-(2,9762 \cdot 10^{-6})(600)}} 0,5 \right] \\
&= 2,20 \times 10^7 \text{ ncm}^{-2} \text{ s}^{-1}
\end{aligned}$$

b. Data cuplikan :

$$\begin{array}{llll}
\text{Cuplikan} & = {}^{197}_{79}\text{Au} & t_i & = 600 \text{ sekon} \\
\rho & = 19,32 \text{ gram cm}^{-3} & t_d & = 697920 \text{ sekon} \\
Bc & = 0,005 \text{ gram} & t_c & = 600 \text{ sekon} \\
\sigma_{act} & = 98,8 \text{ barn} = 9,88 \times 10^{-23} \text{ cm}^2 & C & = 1494,24 \text{ cps} \\
\lambda & = 2,9762 \times 10^{-6} \text{ s}^{-1} & k & = 0,00553 = 5,5310^{-3} \\
N_A & = 6,023 \times 10^{23} \text{ atom/mol} & \Sigma_{act} & = 59,50724 \\
\Delta t_i = \Delta t_d = \Delta t_c & = 0,5 \text{ sekon} & \Delta B & = 0,0005 \text{ gram}
\end{array}$$

$$\Delta\phi = \frac{\lambda^2 C \rho}{k \sum_{act} B_c (1 - e^{-\lambda t_i})(e^{-\lambda t_d})(1 - e^{-\lambda t_c})} \left[ \frac{1}{\lambda Bc} \Delta Bc + \frac{e^{-\lambda t_i}}{1 - e^{-\lambda t_i}} \Delta t_i + \Delta t_d + \frac{e^{-\lambda t_c}}{1 - e^{-\lambda t_c}} \Delta t_c \right]$$

$$= \left[ \frac{(2,9762 \cdot 10^{-6})^2 (1494,24)(19,32)}{(5,53 \cdot 10^{-3})(59,50724)(0,005) \left(1 - e^{-(2,9762 \cdot 10^{-6})(600)}\right) \left(e^{-(2,9762 \cdot 10^{-6})(697920)}\right) \left(1 - e^{-(2,9762 \cdot 10^{-6})(600)}\right)} \right]$$

$$\left[ \frac{1}{((2,9762 \cdot 10^{-6})(0,005))} (0,0005) + \frac{e^{-(2,9762 \cdot 10^{-6})(600)}}{1 - e^{-(2,9762 \cdot 10^{-6})(600)}} 0,5 + 0,5 + \frac{e^{-(2,9762 \cdot 10^{-6})(600)}}{1 - e^{-(2,9762 \cdot 10^{-6})(600)}} 0,5 \right]$$

$$= 1,33 \times 10^7 \text{ ncm}^{-2} \text{ s}^{-1}$$

c. Data cuplikan :

Cuplikan	= ${}^{197}_{79}\text{Au}$	$t_i$	= 600 sekon
$\rho$	= 19,32 gram $\text{cm}^{-3}$	$t_d$	= 706980 sekon
Bc	= 0,006 gram	$t_c$	= 600 sekon
$\sigma_{act}$	= 98,8 barn = $9,88 \times 10^{-23} \text{ cm}^2$	C	= 433,43 cps
$\lambda$	= $2,9762 \times 10^{-6} \text{ s}^{-1}$	k	= $0,00553 = 5,5310^{-3}$
$N_A$	= $6,023 \times 10^{23}$ atom/mol	$\Sigma_{act}$	= 59,50724
$\Delta t_i = \Delta t_d = \Delta t_c$	= 0,5 sekon	$\Delta B$	= 0,0005 gram

$$\Delta\phi = \frac{\lambda^2 C \rho}{k \sum_{act} B_c (1 - e^{-\lambda t_i})(e^{-\lambda t_d})(1 - e^{-\lambda t_c})} \left[ \frac{1}{\lambda Bc} \Delta Bc + \frac{e^{-\lambda t_i}}{1 - e^{-\lambda t_i}} \Delta t_i + \Delta t_d + \frac{e^{-\lambda t_c}}{1 - e^{-\lambda t_c}} \Delta t_c \right]$$

$$= \left[ \frac{(2,9762 \cdot 10^{-6})^2 (433,43)(19,32)}{(5,53 \cdot 10^{-3})(59,50724)(0,006) \left(1 - e^{-(2,9762 \cdot 10^{-6})(600)}\right) \left(e^{-(2,9762 \cdot 10^{-6})(706980)}\right) \left(1 - e^{-(2,9762 \cdot 10^{-6})(600)}\right)} \right]$$

$$\left[ \frac{1}{((2,9762 \cdot 10^{-6})(0,006))} (0,0005) + \frac{e^{-(2,9762 \cdot 10^{-6})(600)}}{1 - e^{-(2,9762 \cdot 10^{-6})(600)}} 0,5 + 0,5 + \frac{e^{-(2,9762 \cdot 10^{-6})(600)}}{1 - e^{-(2,9762 \cdot 10^{-6})(600)}} 0,5 \right]$$

$$= 2,77 \times 10^6 \text{ ncm}^{-2} \text{ s}^{-1}$$

d. Data cuplikan :

Cuplikan	= $^{197}_{79}\text{Au}$	$t_i$	= 600 sekon
$\rho$	= 19,32gram $\text{cm}^{-3}$	$t_d$	= 708300 sekon
Bc	= 0,007 gram	$t_c$	= 600 sekon
$\sigma_{act}$	= 98,8 barn = $9,88 \times 10^{-23} \text{cm}^2$	C	= 220,22cps
$\lambda$	= $2,9762 \times 10^{-6} \text{s}^{-1}$	k	= $0,00553 = 5,5310^{-3}$
$N_A$	= $6,023 \times 10^{23}$ atom/mol	$\Sigma_{act}$	= 59,50724
$\Delta t_i = \Delta t_d = \Delta t_c$	= 0,5 sekon	$\Delta B$	= 0,0005 gram

$$\Delta\phi = \frac{\lambda^2 C \rho}{k \sum_{act} B_c (1 - e^{-\lambda t_i})(e^{-\lambda t_d})(1 - e^{-\lambda t_c})} \left[ \frac{1}{\lambda Bc} \Delta Bc + \frac{e^{-\lambda t_i}}{1 - e^{-\lambda t_i}} \Delta t_i + \Delta t_d + \frac{e^{-\lambda t_c}}{1 - e^{-\lambda t_c}} \Delta t_c \right]$$

$$= \left[ \frac{(2,9762 \cdot 10^{-6})^2 (220,22)(19,32)}{(5,53 \cdot 10^{-3})(59,50724)(0,007) \left(1 - e^{-(2,9762 \cdot 10^{-6})(600)}\right) \left(e^{-(2,9762 \cdot 10^{-6})(708300)}\right) \left(1 - e^{-(2,9762 \cdot 10^{-6})(600)}\right)} \right]$$

$$\left[ \frac{1}{((2,9762 \cdot 10^{-6})(0,007))} (0,0005) + \frac{e^{-(2,9762 \cdot 10^{-6})(600)}}{1 - e^{-(2,9762 \cdot 10^{-6})(600)}} 0,5 + 0,5 + \frac{e^{-(2,9762 \cdot 10^{-6})(600)}}{1 - e^{-(2,9762 \cdot 10^{-6})(600)}} 0,5 \right]$$

$$= 1,04 \times 10^6 \text{ ncm}^{-2} \text{s}^{-1}$$

e. Data cuplikan :

Cuplikan	= $^{197}_{79}\text{Au}$	$t_i$	= 600 sekon
$\rho$	= 19,32gram $\text{cm}^{-3}$	$t_d$	= 709680 sekon
Bc	= 0,009 gram	$t_c$	= 600 sekon
$\sigma_{act}$	= 98,8 barn = $9,88 \times 10^{-23} \text{cm}^2$	C	= 26,54cps
$\lambda$	= $2,9762 \times 10^{-6} \text{s}^{-1}$	k	= $0,00553 = 5,5310^{-3}$



$$N_A = 6,023 \times 10^{23} \text{ atom/mol}$$

$$\Sigma_{act} = 59,50724$$

$$\Delta t_i = \Delta t_d = \Delta t_c = 0,5 \text{ sekon}$$

$$\Delta B = 0,0005 \text{ gram}$$

$$\Delta\phi = \frac{\lambda^2 C \rho}{k \sum_{act} B_c (1 - e^{-\lambda t_i})(e^{-\lambda t_d})(1 - e^{-\lambda t_c})} \left[ \frac{1}{\lambda B_c} \Delta B_c + \frac{e^{-\lambda t_i}}{1 - e^{-\lambda t_i}} \Delta t_i + \Delta t_d + \frac{e^{-\lambda t_c}}{1 - e^{-\lambda t_c}} \Delta t_c \right]$$

$$= \left[ \frac{(2,9762 \cdot 10^{-6})^2 (26,54)(19,32)}{(5,53 \cdot 10^{-3})(59,50724)(0,009) \left(1 - e^{-(2,9762 \cdot 10^{-6})(600)}\right) \left(e^{-(2,9762 \cdot 10^{-6})(709680)}\right) \left(1 - e^{-(2,9762 \cdot 10^{-6})(600)}\right)} \right]$$

$$\left[ \frac{1}{((2,9762 \cdot 10^{-6})(0,009))} (0,0005) + \frac{e^{-(2,9762 \cdot 10^{-6})(600)}}{1 - e^{-(2,9762 \cdot 10^{-6})(600)}} 0,5 + 0,5 + \frac{e^{-(2,9762 \cdot 10^{-6})(600)}}{1 - e^{-(2,9762 \cdot 10^{-6})(600)}} 0,5 \right]$$

$$= 7,66 \times 10^4 \text{ ncm}^{-2} \text{ s}^{-1}$$

## Lampiran 12. Perhitungan Fluks Neutron Thermal

$$\phi_{th} = \phi_{\tau} - \phi_c$$

1. Foil emas 0,004 gram

$$\phi_{\tau} = 3,25 \times 10^{12} \text{ncm}^{-2}\text{s}^{-1}$$

$$\phi_c = 1,06 \times 10^{12} \text{ncm}^{-2}\text{s}^{-1}$$

$$\begin{aligned}\phi_{th} &= (3,25 \times 10^{12}) - (1,06 \times 10^{12}) \text{ncm}^{-2}\text{s}^{-1} \\ &= 2,19 \times 10^{12} \text{ncm}^{-2}\text{s}^{-1}\end{aligned}$$

2. Foil emas 0,005 gram

$$\phi_{\tau} = 2,90 \times 10^{12} \text{ncm}^{-2}\text{s}^{-1}$$

$$\phi_c = 7,99 \times 10^{11} \text{ncm}^{-2}\text{s}^{-1}$$

$$\begin{aligned}\phi_{th} &= (2,90 \times 10^{12}) - (7,99 \times 10^{11}) \text{ncm}^{-2}\text{s}^{-1} \\ &= 2,10 \times 10^{12} \text{ncm}^{-2}\text{s}^{-1}\end{aligned}$$

3. Foil emas 0,006 gram

$$\phi_{\tau} = 1,56 \times 10^{12} \text{ncm}^{-2}\text{s}^{-1}$$

$$\phi_c = 1,99 \times 10^{11} \text{ncm}^{-2}\text{s}^{-1}$$

$$\begin{aligned}\phi_{th} &= (1,56 \times 10^{12}) - (1,99 \times 10^{11}) \text{ncm}^{-2}\text{s}^{-1} \\ &= 1,36 \times 10^{12} \text{ncm}^{-2}\text{s}^{-1}\end{aligned}$$

4. Foil emas 0,007 gram

$$\phi_{\tau} = 8,69 \times 10^{11} \text{ncm}^{-2}\text{s}^{-1}$$

$$\phi_c = 8,71 \times 10^{10} \text{ncm}^{-2}\text{s}^{-1}$$

$$\begin{aligned}\phi_{th} &= (8,69 \times 10^{11}) - (8,71 \times 10^{10}) \text{ ncm}^{-2}\text{s}^{-1} \\ &= 7,82 \times 10^{11} \text{ ncm}^{-2}\text{s}^{-1}\end{aligned}$$

5. Foil emas 0,009 gram

$$\phi_{\tau} = 1,80 \times 10^{11} \text{ ncm}^{-2}\text{s}^{-1}$$

$$\phi_c = 8,19 \times 10^9 \text{ ncm}^{-2}\text{s}^{-1}$$

$$\begin{aligned}\phi_{th} &= (1,80 \times 10^{11}) - (8,19 \times 10^9) \text{ ncm}^{-2}\text{s}^{-1} \\ &= 1,72 \times 10^{11} \text{ ncm}^{-2}\text{s}^{-1}\end{aligned}$$

Dari hasil-hasil fluks thermal di atas maka dapat dihitung fluks neutron thermal sebagai berikut :

$$\begin{aligned}\phi_{th} &= \frac{\phi_{t1} + \phi_{t2} + \phi_{t3} + \phi_{t4} + \phi_{t5}}{5} \\ \phi_{th} &= \frac{(2,19 \times 10^{12}) + (2,10 \times 10^{12}) + (1,36 \times 10^{12}) + (7,82 \times 10^{11}) + (1,72 \times 10^{11})}{5} \text{ ncm}^{-2}\text{s}^{-1} \\ \phi_{th} &= 1,32 \times 10^{12} \text{ ncm}^{-2}\text{s}^{-1}\end{aligned}$$

### Lampiran 13. Nilai Ralat Fluks Neutron Thermal

$$\Delta\phi_{th} = \Delta\phi_{\tau} + \Delta\phi_c$$

1. Foil emas 0,004 gram

$$\Delta\phi_{\tau} = 6,73 \times 10^7 \text{ ncm}^{-2} \text{ s}^{-1}$$

$$\Delta\phi_c = 2,20 \times 10^7 \text{ ncm}^{-2} \text{ s}^{-1}$$

$$\begin{aligned} \Delta\phi_{th} &= (6,73 \times 10^7) + (2,20 \times 10^7) \text{ ncm}^{-2} \text{ s}^{-1} \\ &= 8,93 \times 10^7 \text{ ncm}^{-2} \text{ s}^{-1} \end{aligned}$$

2. Foil emas 0,005 gram

$$\Delta\phi_{\tau} = 4,82 \times 10^7 \text{ ncm}^{-2} \text{ s}^{-1}$$

$$\Delta\phi_c = 1,33 \times 10^7 \text{ ncm}^{-2} \text{ s}^{-1}$$

$$\begin{aligned} \Delta\phi_{th} &= (4,82 \times 10^7) + (1,33 \times 10^7) \text{ ncm}^{-2} \text{ s}^{-1} \\ &= 6,15 \times 10^7 \text{ ncm}^{-2} \text{ s}^{-1} \end{aligned}$$

3. Foil emas 0,006 gram

$$\Delta\phi_{\tau} = 2,16 \times 10^7 \text{ ncm}^{-2} \text{ s}^{-1}$$

$$\Delta\phi_c = 2,77 \times 10^6 \text{ ncm}^{-2} \text{ s}^{-1}$$

$$\begin{aligned} \Delta\phi_{th} &= (2,16 \times 10^7) + (2,77 \times 10^6) \text{ ncm}^{-2} \text{ s}^{-1} \\ &= 2,44 \times 10^7 \text{ ncm}^{-2} \text{ s}^{-1} \end{aligned}$$

4. Foil emas 0,007 gram

$$\Delta\phi_{\tau} = 1,04 \times 10^7 \text{ ncm}^{-2} \text{ s}^{-1}$$

$$\Delta\phi_c = 1,04 \times 10^6 \text{ ncm}^{-2} \text{ s}^{-1}$$

$$\begin{aligned}\Delta\phi_{th} &= (1,04 \times 10^7) + (1,04 \times 10^6) \text{ ncm}^{-2}\text{s}^{-1} \\ &= 1,14 \times 10^7 \text{ ncm}^{-2}\text{s}^{-1}\end{aligned}$$

5. Foil emas 0,009 gram

$$\begin{aligned}\Delta\phi_{\tau} &= 1,68 \times 10^6 \text{ ncm}^{-2}\text{s}^{-1} \\ \Delta\phi_c &= 7,66 \times 10^4 \text{ ncm}^{-2}\text{s}^{-1} \\ \Delta\phi_{th} &= (1,68 \times 10^6) + (7,66 \times 10^4) \text{ ncm}^{-2}\text{s}^{-1} \\ &= 1,76 \times 10^6 \text{ ncm}^{-2}\text{s}^{-1}\end{aligned}$$

Dari hasil-hasil nilai ralat fluks thermal di atas maka dapat dihitung rata-rata nilai ralat fluks neutron thermal sebagai berikut :

$$\begin{aligned}\Delta\phi_{th} &= \frac{\Delta\phi_{t1} + \Delta\phi_{t2} + \Delta\phi_{t3} + \Delta\phi_{t4} + \Delta\phi_{t5}}{5} \\ \Delta\phi_{th} &= \frac{(8,93 \times 10^7) + (6,15 \times 10^7) + (2,44 \times 10^7) + (1,14 \times 10^7) + (1,76 \times 10^6)}{5} \text{ ncm}^{-2}\text{s}^{-1} \\ \Delta\phi_{th} &= 3,77 \times 10^7 \text{ ncm}^{-2}\text{s}^{-1}\end{aligned}$$

## Lampiran 14. Nilai Banding Cadmium

$$R_{cd} = \frac{\phi_{\tau}}{\phi_c}$$

1. Foil emas 0,004 gram

$$\phi_{\tau} = 3,25 \times 10^{12} \text{ncm}^{-2} \text{s}^{-1}$$

$$\phi_c = 1,06 \times 10^{12} \text{ncm}^{-2} \text{s}^{-1}$$

$$R_{cd} = \frac{3,25 \times 10^{12} \text{ncm}^{-2} \text{s}^{-1}}{1,06 \times 10^{12} \text{ncm}^{-2} \text{s}^{-1}}$$

$$R_{cd} = 3,06$$

2. Foil emas 0,005 gram

$$\phi_{\tau} = 2,90 \times 10^{12} \text{ncm}^{-2} \text{s}^{-1}$$

$$\phi_c = 7,99 \times 10^{11} \text{ncm}^{-2} \text{s}^{-1}$$

$$R_{cd} = \frac{2,90 \times 10^{12} \text{ncm}^{-2} \text{s}^{-1}}{7,99 \times 10^{11} \text{ncm}^{-2} \text{s}^{-1}}$$

$$R_{cd} = 3,63$$

3. Foil emas 0,006 gram

$$\phi_{\tau} = 1,56 \times 10^{12} \text{ncm}^{-2} \text{s}^{-1}$$

$$\phi_c = 1,99 \times 10^{11} \text{ncm}^{-2} \text{s}^{-1}$$

$$R_{cd} = \frac{1,56 \times 10^{12} \text{ncm}^{-2} \text{s}^{-1}}{1,99 \times 10^{11} \text{ncm}^{-2} \text{s}^{-1}}$$

$$R_{cd} = 7,85$$

4. Foil emas 0,007 gram

$$\phi_{\tau} = 8,69 \times 10^{11} \text{ncm}^{-2} \text{s}^{-1}$$

$$\phi_{\text{c}} = 8,71 \times 10^{10} \text{ncm}^{-2} \text{s}^{-1}$$

$$R_{\text{cd}} = \frac{8,69 \times 10^{11} \text{ncm}^{-2} \text{s}^{-1}}{8,71 \times 10^{10} \text{ncm}^{-2} \text{s}^{-1}}$$

$$R_{\text{cd}} = 9,99$$

5. Foil emas 0,009 gram

$$\phi_{\tau} = 1,80 \times 10^{11} \text{ncm}^{-2} \text{s}^{-1}$$

$$\phi_{\text{c}} = 8,19 \times 10^9 \text{ncm}^{-2} \text{s}^{-1}$$

$$R_{\text{cd}} = \frac{1,80 \times 10^{11} \text{ncm}^{-2} \text{s}^{-1}}{8,19 \times 10^9 \text{ncm}^{-2} \text{s}^{-1}}$$

$$R_{\text{cd}} = 21,98$$

Dari hasil-hasil nilai banding cadmium di atas maka dapat dihitung rata-rata nilai banding cadmium sebagai berikut :

$$R_{\text{cd}} = \frac{R_{\text{cd}1} + R_{\text{cd}2} + R_{\text{cd}3} + R_{\text{cd}4} + R_{\text{cd}5}}{5}$$

$$= \frac{3,06 + 3,63 + 7,85 + 9,99 + 21,98}{5}$$

$$= 9,30$$

## Lampiran 15. Ralat Nilai Banding Cadmium

$$\Delta R_{cd} = \left| \frac{1}{\phi_c} \right| \Delta \phi_\tau + \left| \frac{\phi_\tau}{\phi_c^2} \right| \Delta \phi_c$$

1. Foil emas 0,004 gram

$$\phi_\tau = 3,25 \times 10^{12} \text{ncm}^{-2} \text{s}^{-1}$$

$$\phi_c = 1,06 \times 10^{12} \text{ncm}^{-2} \text{s}^{-1}$$

$$\Delta \phi_\tau = 6,73 \times 10^7 \text{ncm}^{-2} \text{s}^{-1}$$

$$\Delta \phi_c = 2,20 \times 10^7 \text{ncm}^{-2} \text{s}^{-1}$$

$$\Delta R_{cd} = \left| \frac{1}{1,06 \times 10^{12} \text{ncm}^{-2} \text{s}^{-1}} \right| (6,73 \times 10^7 \text{ncm}^{-2} \text{s}^{-1}) + \left| \frac{3,25 \times 10^{12} \text{ncm}^{-2} \text{s}^{-1}}{(1,06 \times 10^{12} \text{ncm}^{-2} \text{s}^{-1})^2} \right| (2,20 \times 10^7 \text{ncm}^{-2} \text{s}^{-1})$$

$$\Delta R_{cd} = 1,27 \times 10^{-4}$$

2. Foil emas 0,005 gram

$$\phi_\tau = 2,90 \times 10^{12} \text{ncm}^{-2} \text{s}^{-1}$$

$$\phi_c = 7,99 \times 10^{11} \text{ncm}^{-2} \text{s}^{-1}$$

$$\Delta \phi_\tau = 4,82 \times 10^7 \text{ncm}^{-2} \text{s}^{-1}$$

$$\Delta \phi_c = 1,33 \times 10^7 \text{ncm}^{-2} \text{s}^{-1}$$

$$\Delta R_{cd} = \left| \frac{1}{7,99 \times 10^{11} \text{ncm}^{-2} \text{s}^{-1}} \right| (4,82 \times 10^7 \text{ncm}^{-2} \text{s}^{-1}) + \left| \frac{2,90 \times 10^{12} \text{ncm}^{-2} \text{s}^{-1}}{(7,99 \times 10^{11} \text{ncm}^{-2} \text{s}^{-1})^2} \right| (1,33 \times 10^7 \text{ncm}^{-2} \text{s}^{-1})$$

$$\Delta R_{cd} = 1,21 \times 10^{-4}$$

3. Foil emas 0,006 gram

$$\phi_\tau = 1,56 \times 10^{12} \text{ncm}^{-2} \text{s}^{-1}$$



$$\phi_c = 1,99 \times 10^{11} \text{ncm}^{-2}\text{s}^{-1}$$

$$\Delta\phi_\tau = 2,16 \times 10^7 \text{ncm}^{-2}\text{s}^{-1}$$

$$\Delta\phi_c = 2,77 \times 10^6 \text{ncm}^{-2}\text{s}^{-1}$$

$$\Delta R_{cd} = \left| \frac{1}{1,99 \times 10^{11} \text{ncm}^{-2}\text{s}^{-1}} \right| (2,16 \times 10^7 \text{ncm}^{-2}\text{s}^{-1}) + \left| \frac{1,56 \times 10^{12} \text{ncm}^{-2}\text{s}^{-1}}{(1,99 \times 10^{11} \text{ncm}^{-2}\text{s}^{-1})^2} \right| (2,77 \times 10^6 \text{ncm}^{-2}\text{s}^{-1})$$

$$\Delta R_{cd} = 2,18 \times 10^{-4}$$

4. Foil emas 0,007 gram

$$\phi_\tau = 8,69 \times 10^{11} \text{ncm}^{-2}\text{s}^{-1}$$

$$\phi_c = 8,71 \times 10^{10} \text{ncm}^{-2}\text{s}^{-1}$$

$$\Delta\phi_\tau = 1,04 \times 10^7 \text{ncm}^{-2}\text{s}^{-1}$$

$$\Delta\phi_c = 1,04 \times 10^6 \text{ncm}^{-2}\text{s}^{-1}$$

$$\Delta R_{cd} = \left| \frac{1}{8,71 \times 10^{10} \text{ncm}^{-2}\text{s}^{-1}} \right| (1,04 \times 10^7 \text{ncm}^{-2}\text{s}^{-1}) + \left| \frac{8,69 \times 10^{11} \text{ncm}^{-2}\text{s}^{-1}}{(8,71 \times 10^{10} \text{ncm}^{-2}\text{s}^{-1})^2} \right| (1,04 \times 10^6 \text{ncm}^{-2}\text{s}^{-1})$$

$$\Delta R_{cd} = 2,39 \times 10^{-4}$$

5. Foil emas 0,009 gram

$$\phi_\tau = 1,80 \times 10^{11} \text{ncm}^{-2}\text{s}^{-1}$$

$$\phi_c = 8,19 \times 10^9 \text{ncm}^{-2}\text{s}^{-1}$$

$$\Delta\phi_\tau = 1,68 \times 10^6 \text{ncm}^{-2}\text{s}^{-1}$$

$$\Delta\phi_c = 7,66 \times 10^4 \text{ncm}^{-2}\text{s}^{-1}$$

$$\Delta R_{cd} = \left| \frac{1}{8,19 \times 10^9 \text{ncm}^{-2}\text{s}^{-1}} \right| (1,68 \times 10^6 \text{ncm}^{-2}\text{s}^{-1}) + \left| \frac{1,80 \times 10^{11} \text{ncm}^{-2}\text{s}^{-1}}{(8,19 \times 10^9 \text{ncm}^{-2}\text{s}^{-1})^2} \right| (7,66 \times 10^4 \text{ncm}^{-2}\text{s}^{-1})$$

$$\Delta R_{cd} = 4,11 \times 10^{-4}$$

Dari hasil-hasil nilai ralat nilai banding cadmium di atas maka dapat dihitung rata-rata nilai ralat nilai banding cadmium sebagai berikut :

$$\Delta R_{cd} = \frac{R_{cd1} + R_{cd2} + R_{cd3} + R_{cd4} + R_{cd5}}{5}$$

$$\Delta R_{cd} = \frac{1,27 \times 10^{-4} + 1,21 \times 10^{-4} + 2,18 \times 10^{-4} + 2,39 \times 10^{-4} + 4,11 \times 10^{-4}}{5}$$

$$\Delta R_{cd} = 2,23 \times 10^{-4}$$